



Modelling of Acid Extraction into TODGA / Octanol Mixtures

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- i-SANEX Solvent
 - 0.2M TODGA
 - 5v/v% Octanol
 - Inert diluent (e.g. OK or TPH)
- Main Extractant (TODGA) extracts Actinides / Lanthanides
 - Also extracts acid (HNO_3)
 - Does not extract most fission products
- Phase Modifier (Octanol) supresses 3rd Phase formation
 - Extracts Acid – but little else

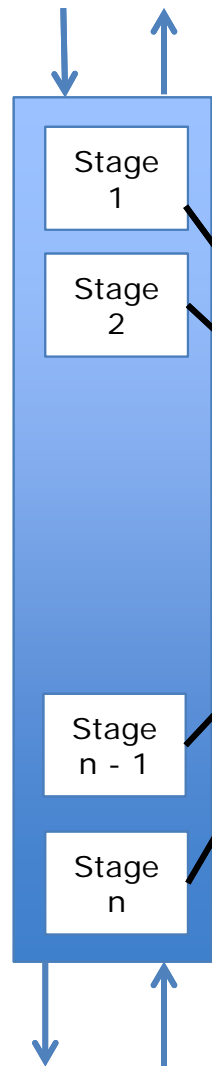


- Production of dynamic process model of proposed flowsheets for:
 - Process Optimisation
 - Sensitivity studies allow many permutations to be analysed
 - Identify most useful experiments to run.
 - Support Safety Analysis
 - Maloperation Studies – What if?
 - Stability studies – flowsheet might look good but how wide is operating envelope?
 - Requirement for above is model that is valid over a wide range of operating conditions
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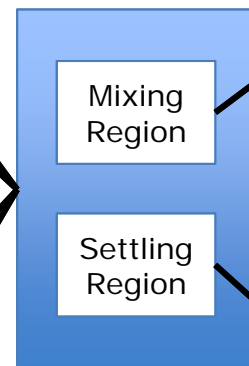
Elements of a Process Model

Flowsheet

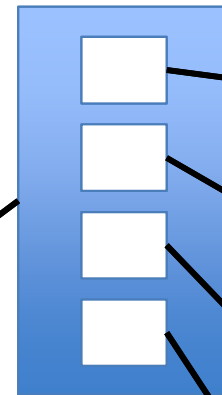
Feeds, Products



Contactor



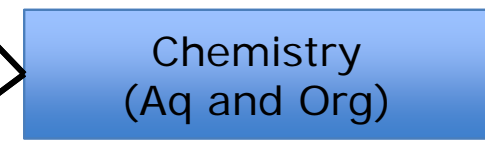
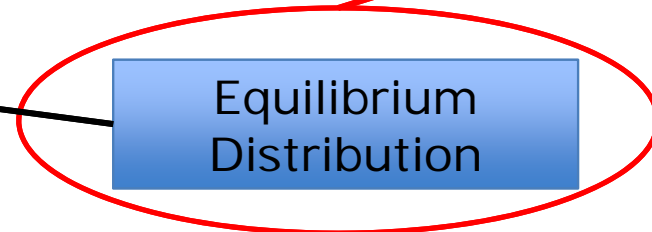
Mixing Region



Settling Region



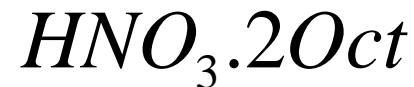
This Talk



Applies Mixer-Settlers and Centrifugal Contactors – Pulsed Columns Different

D-Value Calculations (1)

- Various Solvent phase complexes formed.
- Each solvent phase complex considered separately.
- e.g. acid extracting into octanol includes



- Associated equilibrium: -

$$K = \frac{[HNO_3 \cdot 2Oct]_{(org)}}{\gamma_{H^+} [H^+]_{(aq)} \gamma_{NO_3^-} [NO_3^-]_{(aq)} [Oct_{Free}]_{(org)}^2}$$

- Need to fit expressions for γ 's and value for K

D-Value Calculations (2)

- Data does not allow us to distinguish $\gamma_{H^+}, \gamma_{NO_3^-}$
- Can use γ_{\pm} where $\gamma_{\pm}^2 = \gamma_{H^+} \gamma_{NO_3^-}$
- γ_{\pm} available from literature – but as used here includes allowance for solvent phase activity effects.
- Expression for γ_{\pm} is therefore ultimately empirical
- Aim to same expression for γ_{\pm} in all equilibria
- In general K 's and γ_{\pm} dependent on T – current data all insufficient to determine T dependence

- Composition

- 0.2M TODGA
- 5 v/v% Octanol
- Inert Diluent

- Approach

1. Assume the diluent is genuinely inert
2. Derive separate expressions for acid extraction into TODGA and Octanol
3. Investigate synergistic and antagonistic effects



Acid Extraction into Octanol (1)

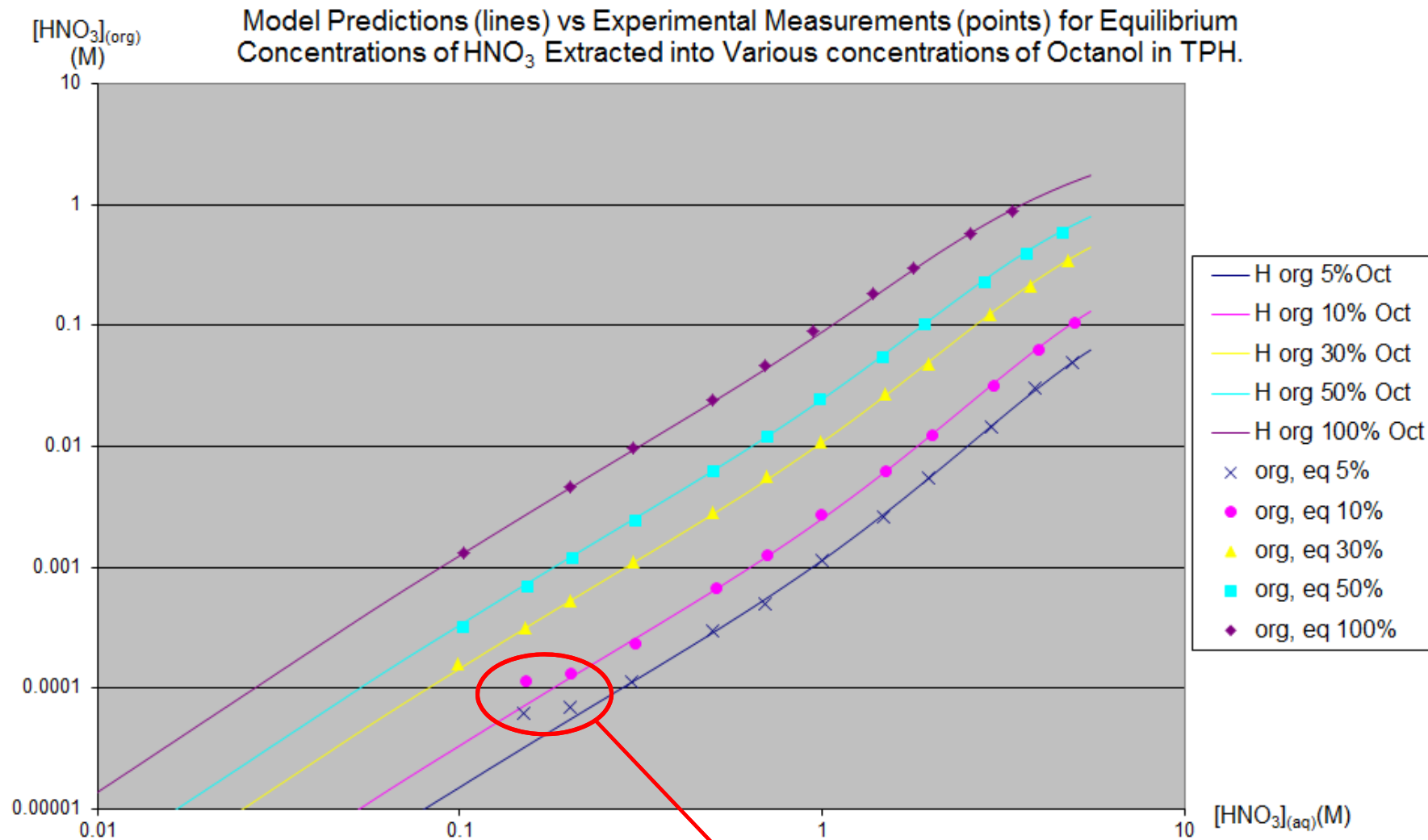
- Data from KIT-INE, 5 – 100% Octanol, 0 – 5M HNO₃
- Previous modelling used HNO₃.2Oct complex only [1]. Good fit but required conditional equilibrium constants – difficult to use in model.

- New model assumes additional HNO₃.Oct, HNO₃.3Oct complexes.

$$[HNO_3]_{org} = \sum_{i=1}^{i=3} \beta_i \gamma_{\pm}^2 [H^+] [NO_3^-] [Oct_{free}]^i$$
$$\gamma_{\pm} = \frac{A}{([NO_3^-] + B)^2} + C + D[NO_3^-] + E[NO_3^-]^2$$

- β_i 's and A through E fitted constants. Form of 2nd equation gives good fit to published activity coefficient data.
- No additional benefit found from adding 2HNO₃.Oct or HNO₃.4Oct complexes

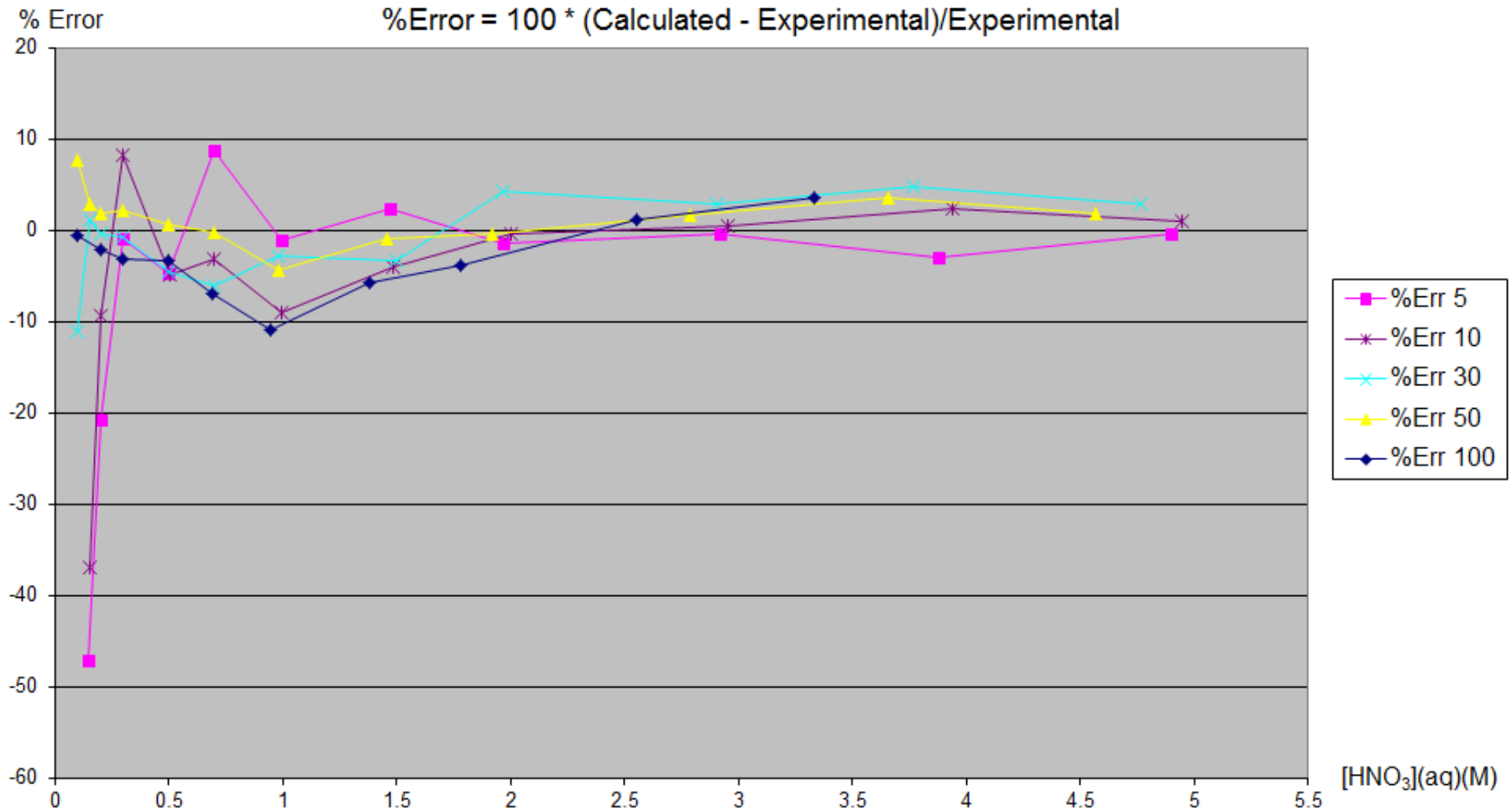
Acid Extraction into Octanol (2)



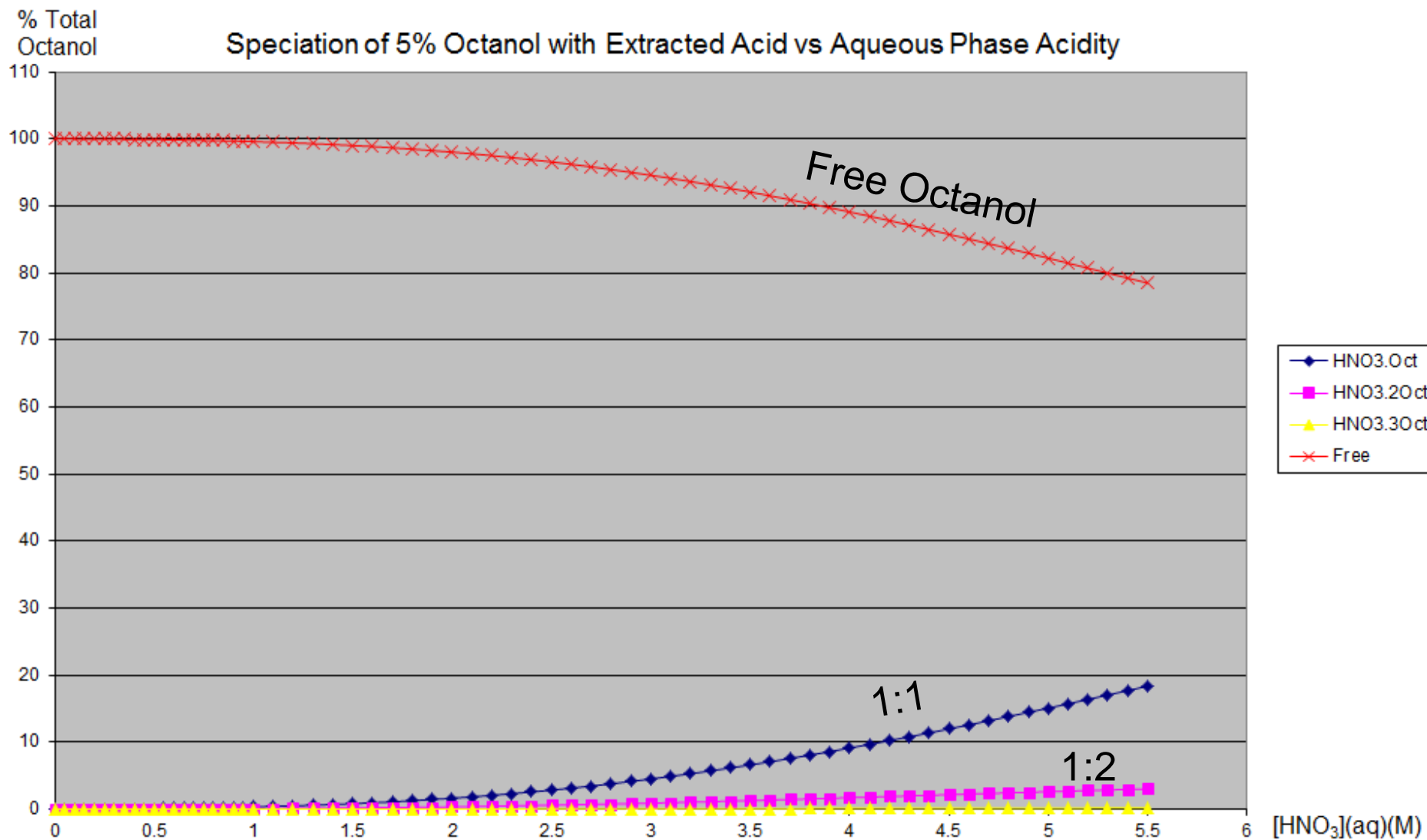
Very good fit except at lowest acidity

Acid Extraction into Octanol (3)

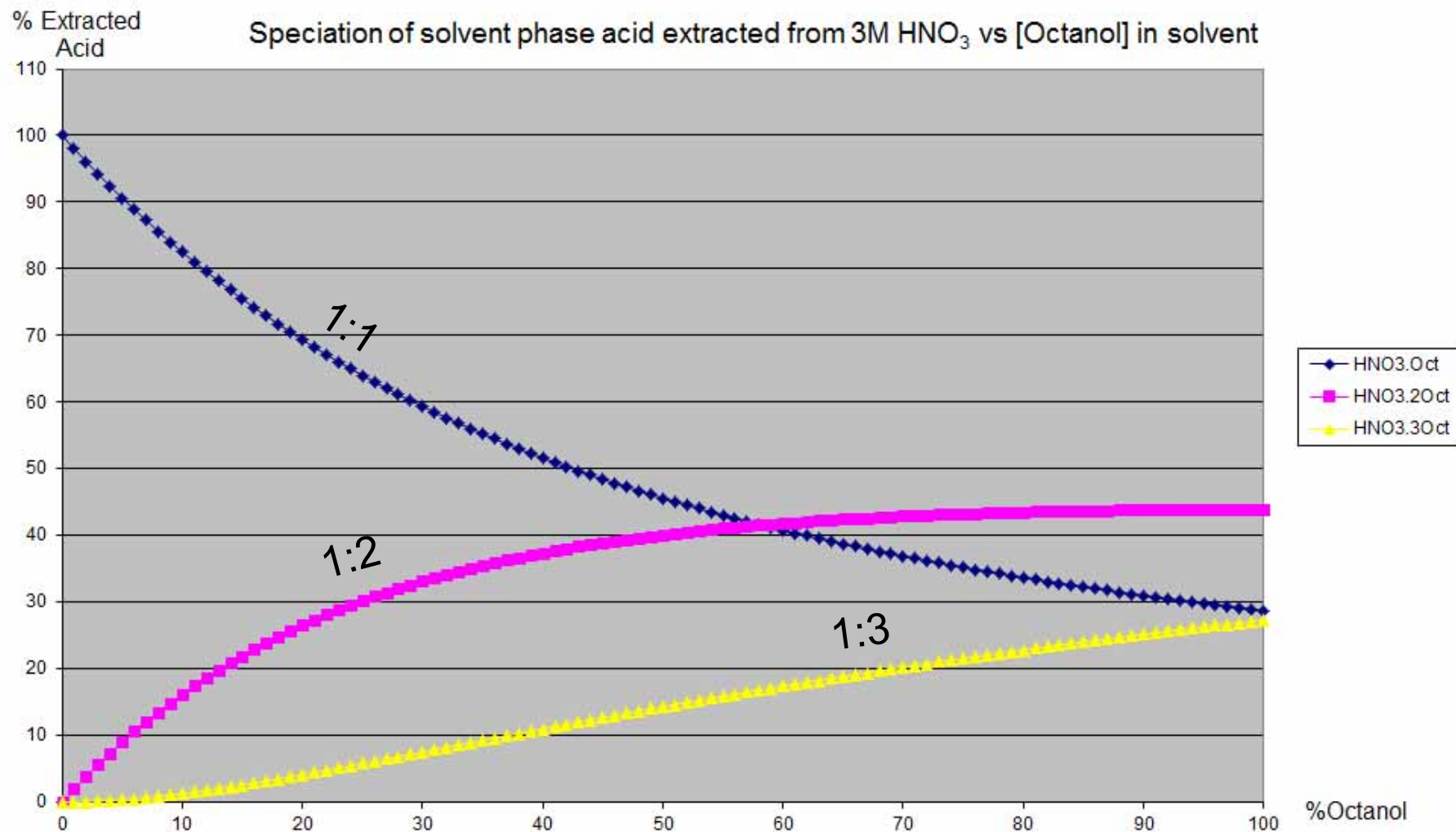
Error in Prediction of Equilibrium Concentrations of HNO_3 Extracted into Various Concentrations of Octanol in TPH.



Acid Extraction into Octanol (4)



Acid Extraction into Octanol (5)



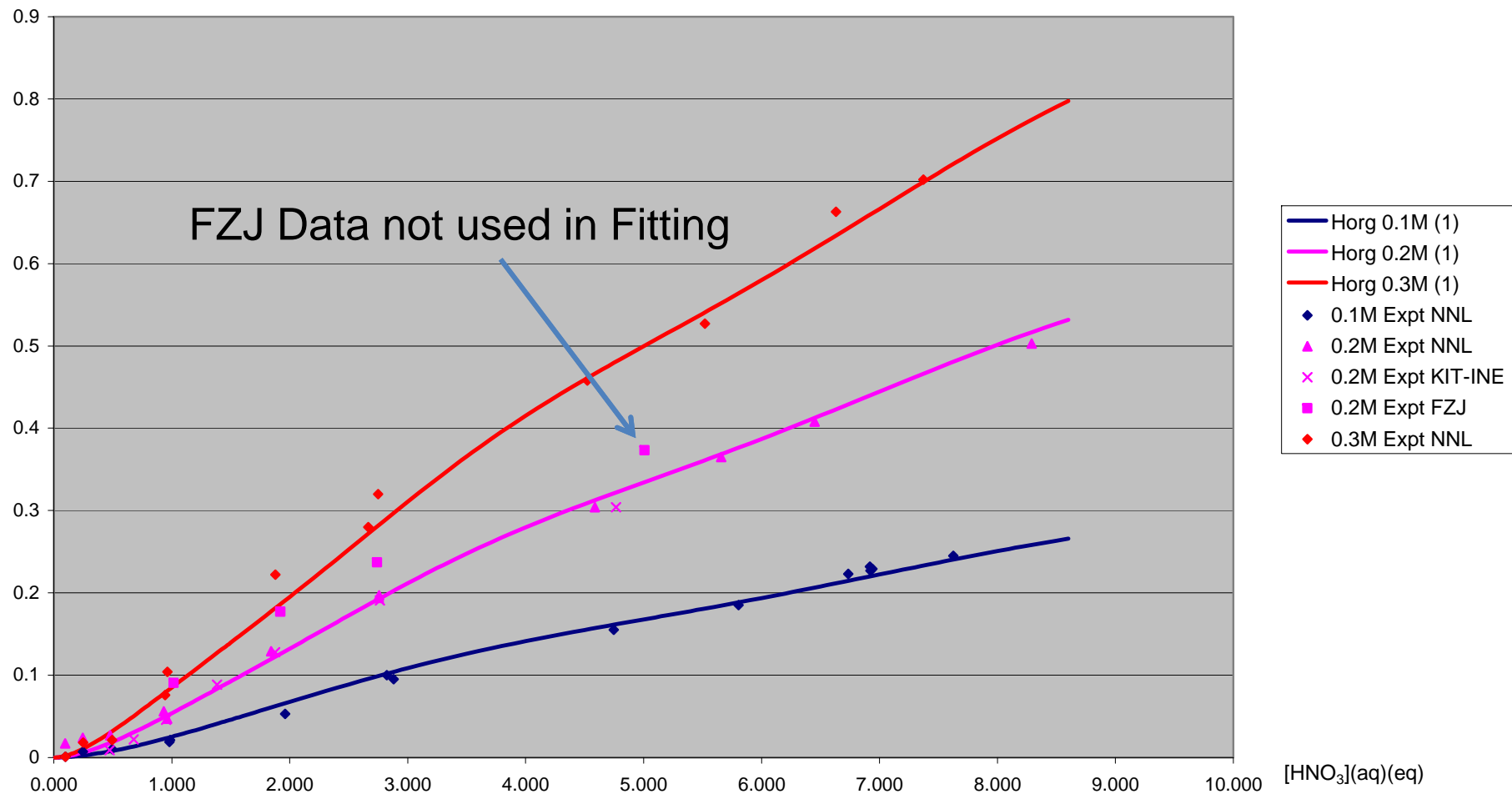
Octanol : Acid ratio in complexes increases with increasing [octanol]

Acid Extraction into TODGA (1)

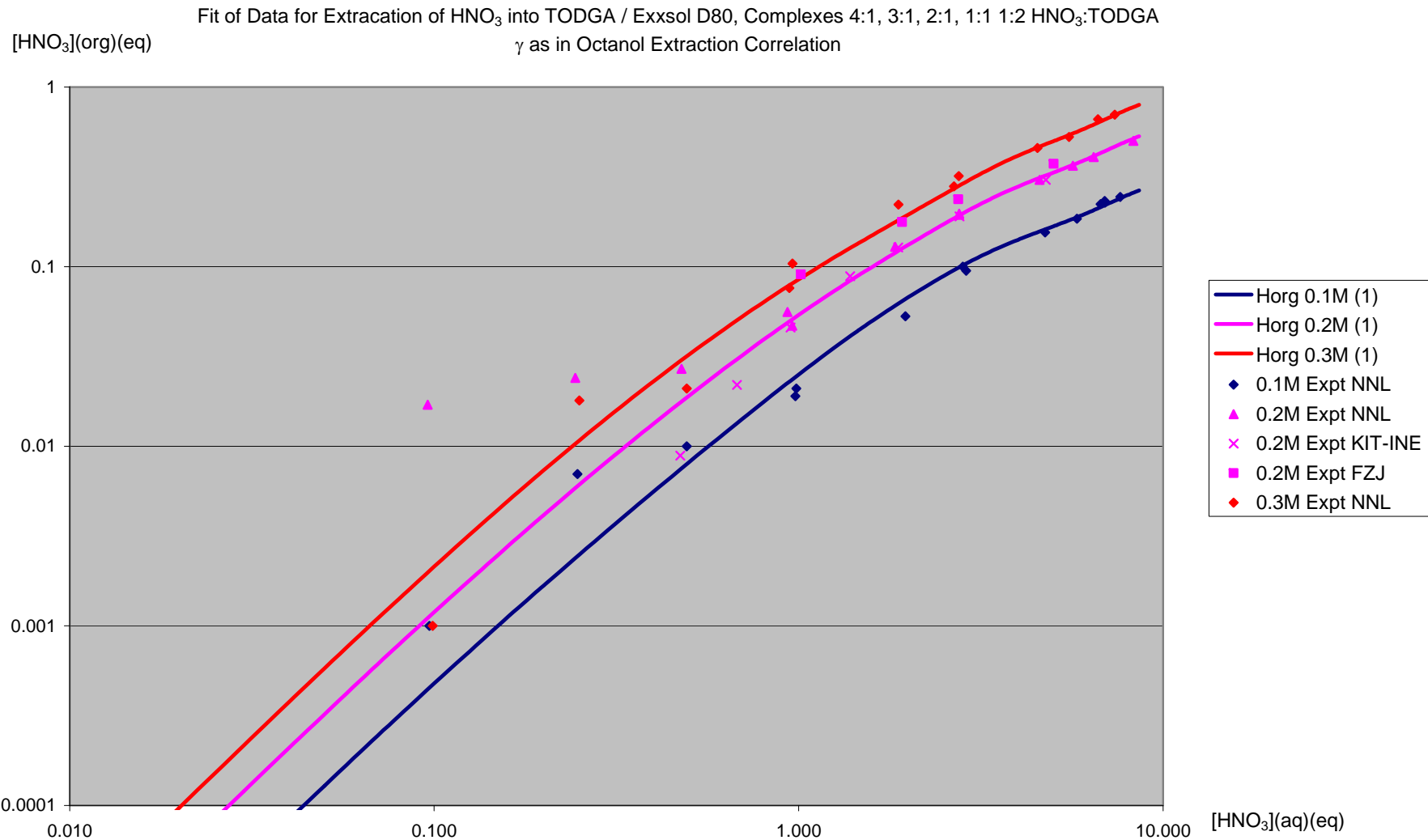
- Data from NNL, KIT-INE and FZJ. 0.05 – 0.4M TODGA, 0 – 8M HNO₃
- Low acidity data widely scattered
- Fitting assumed HNO₃.2TODGA, nHNO₃.TODGA (n = 1, 2, 3, 4)
- Fit set 3HNO₃.TODGA at zero.
- Also tried refitting constants in expression for γ
 - Little benefit found over using same γ expression as for acid extraction into octanol correlation
 - Practically and theoretically attractive to use same γ throughout
- Also attempted including TODGA dimer
 - No benefit from this found

Acid Extraction into TODGA(2)

[HNO₃](org)(eq) Fit of Data for Extracation of HNO₃ into TODGA / Exxsol D80, Complexes 4:1, 3:1, 2:1, 1:1 1:2 HNO₃:TODGA
γ as in Octanol Extraction Correlation



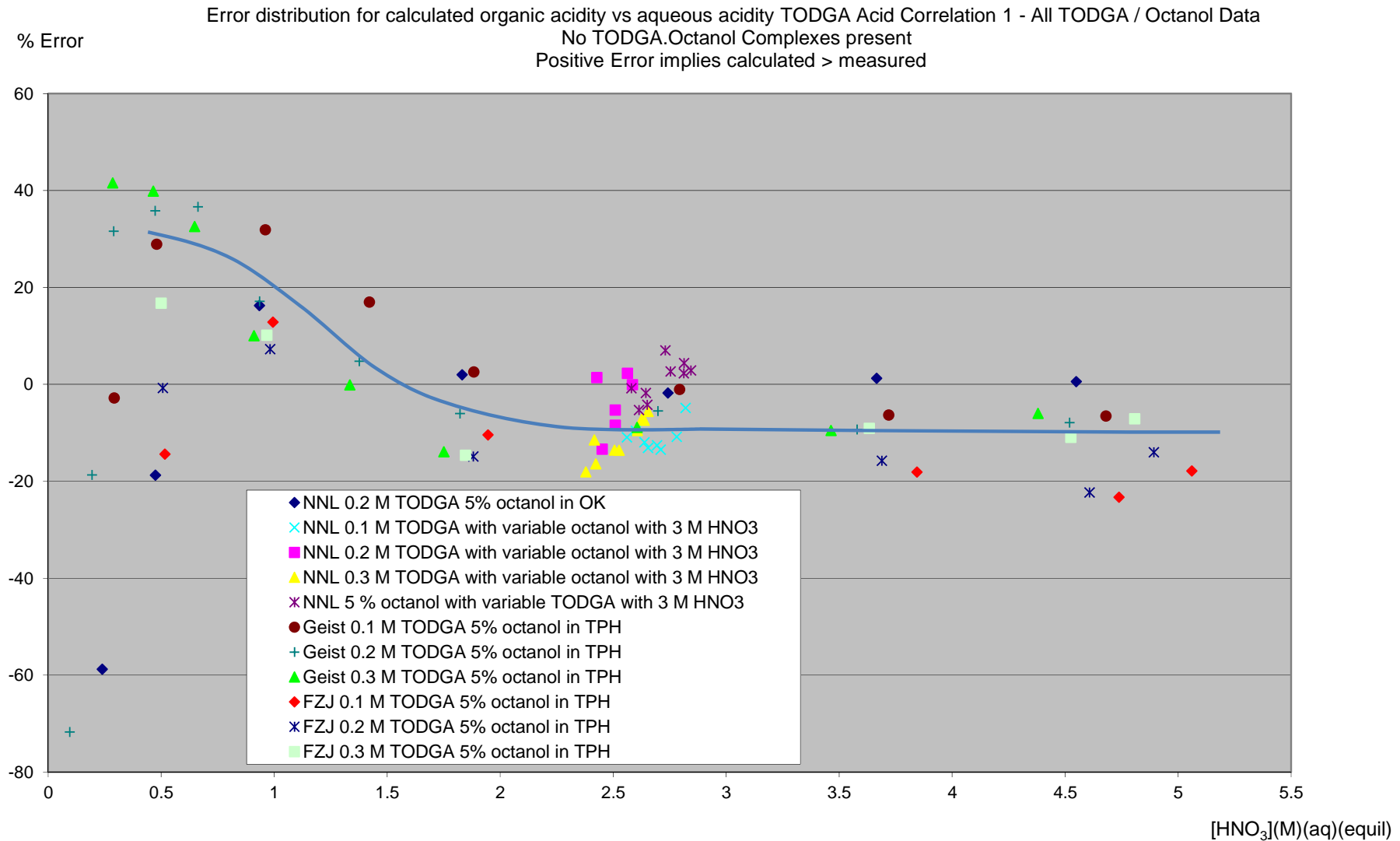
Acid Extraction into TODGA(3)



Low Acid Data Widely Scattered – Difficult to fit well, reduced weights used in fit

- Data from NNL, KIT-INE and FZJ:
 - 0.05 – 0.4M TODGA, 5 – 50v/v% Octanol, 0 – 5M HNO₃
 - Majority at 5% Octanol
- Significant scatter in low acidity data
- Initially assume that TODGA and Octanol do not interact and extract acid independently of each other. i.e. No synergism or antagonism.
- Combined model:
 - Overpredicts for 0.5 → 1.5M HNO₃
 - Underpredicts for [HNO₃] > 1.5M
 - Data for below 0.5M [HNO₃] too scattered to comment meaningfully

Mixed TODGA / Octanol Systems (2)



Superposition of Algorithms for extraction of Acid into TODGA and Octanol

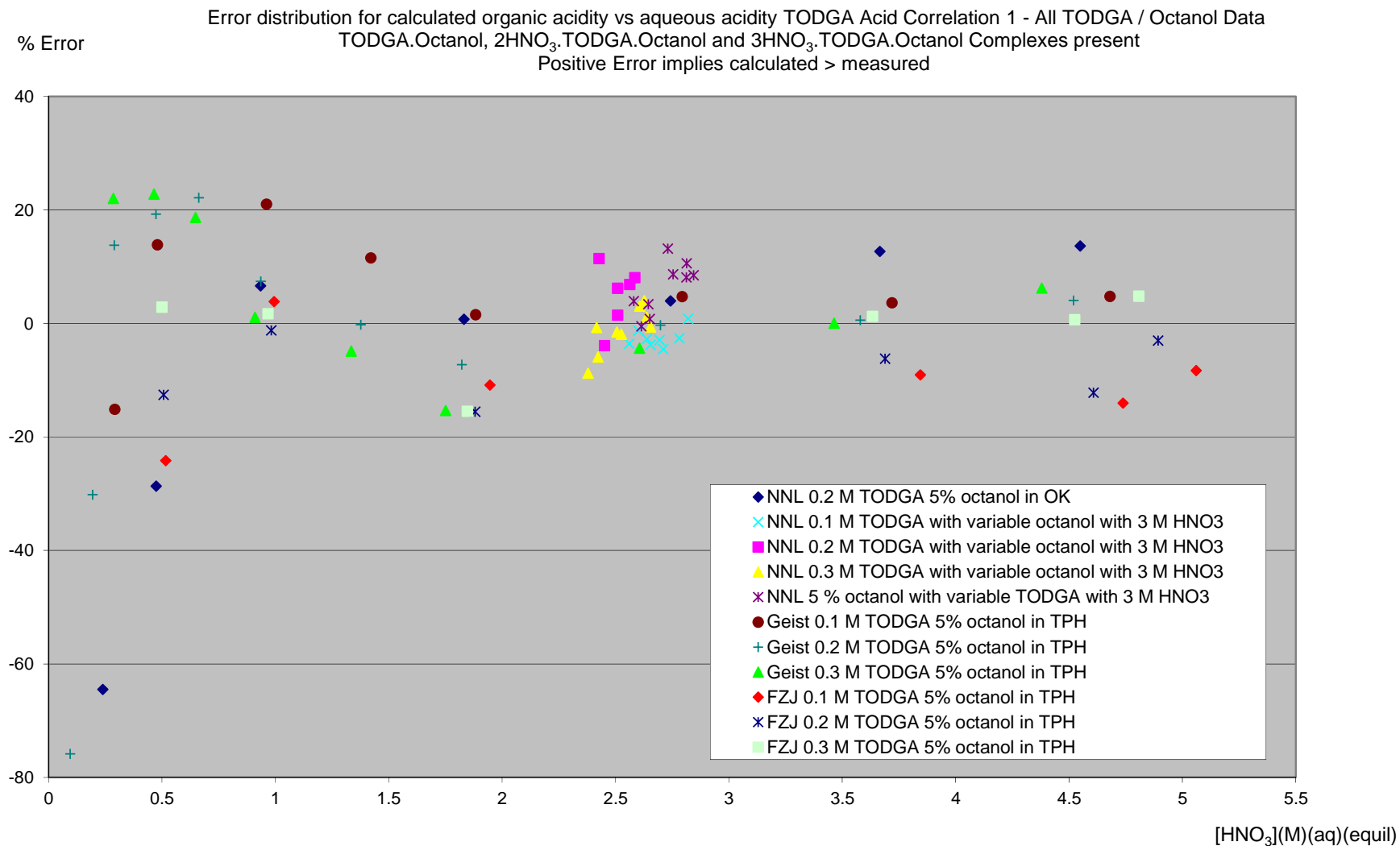
- Suggestion of some Synergistic (higher acidity) and Antagonistic (lower acidity) effects.
- New model developed and fitted with complexes of form



- $n = 0$ complex gives antagonism, others synergism
- Fitting always resulted in elimination of the $n = 1$ complex
- New model can largely remove biases for $[\text{HNO}_3] > 0.5\text{M}$

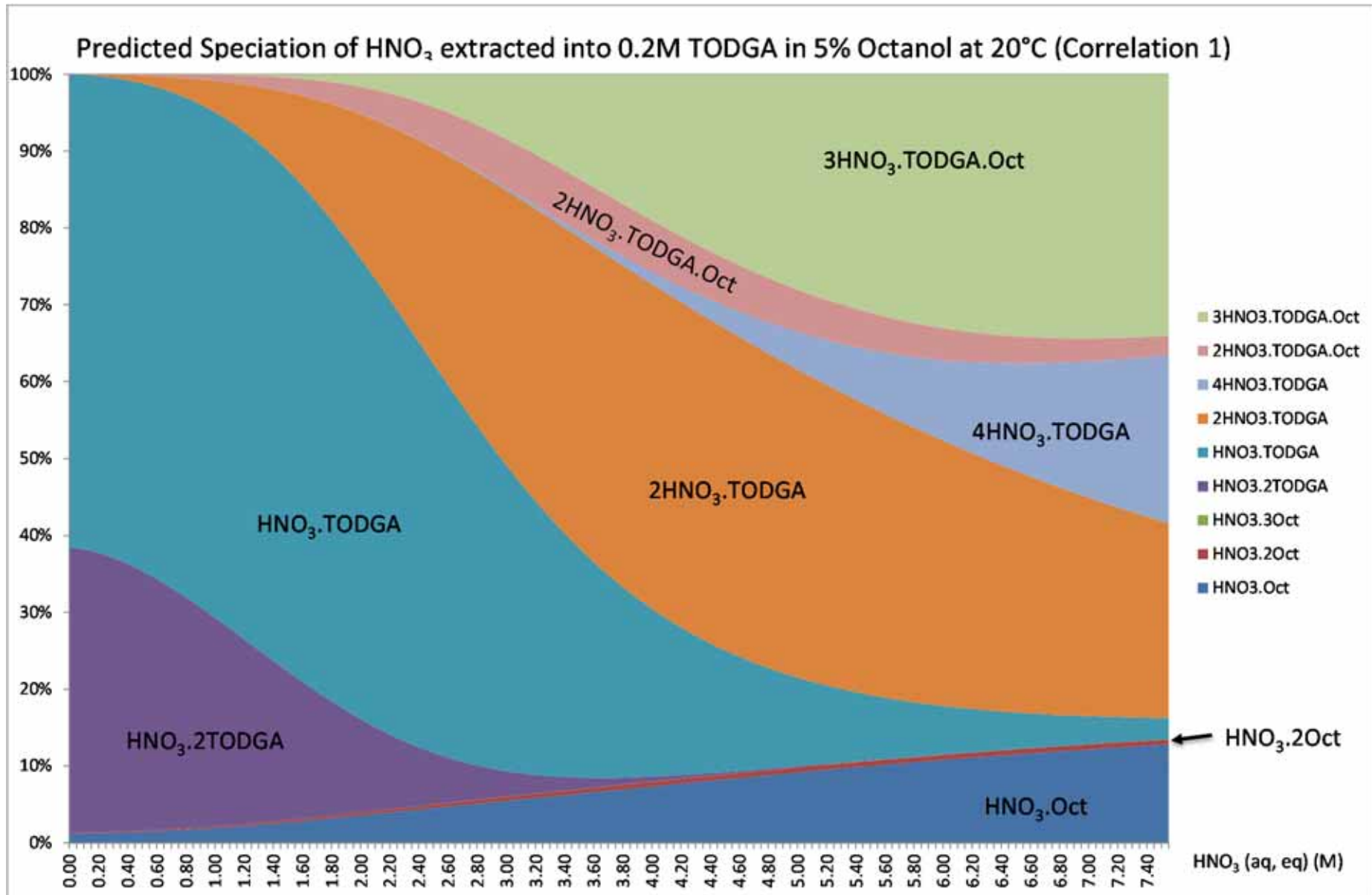


Mixed TODGA / Octanol Systems (4)

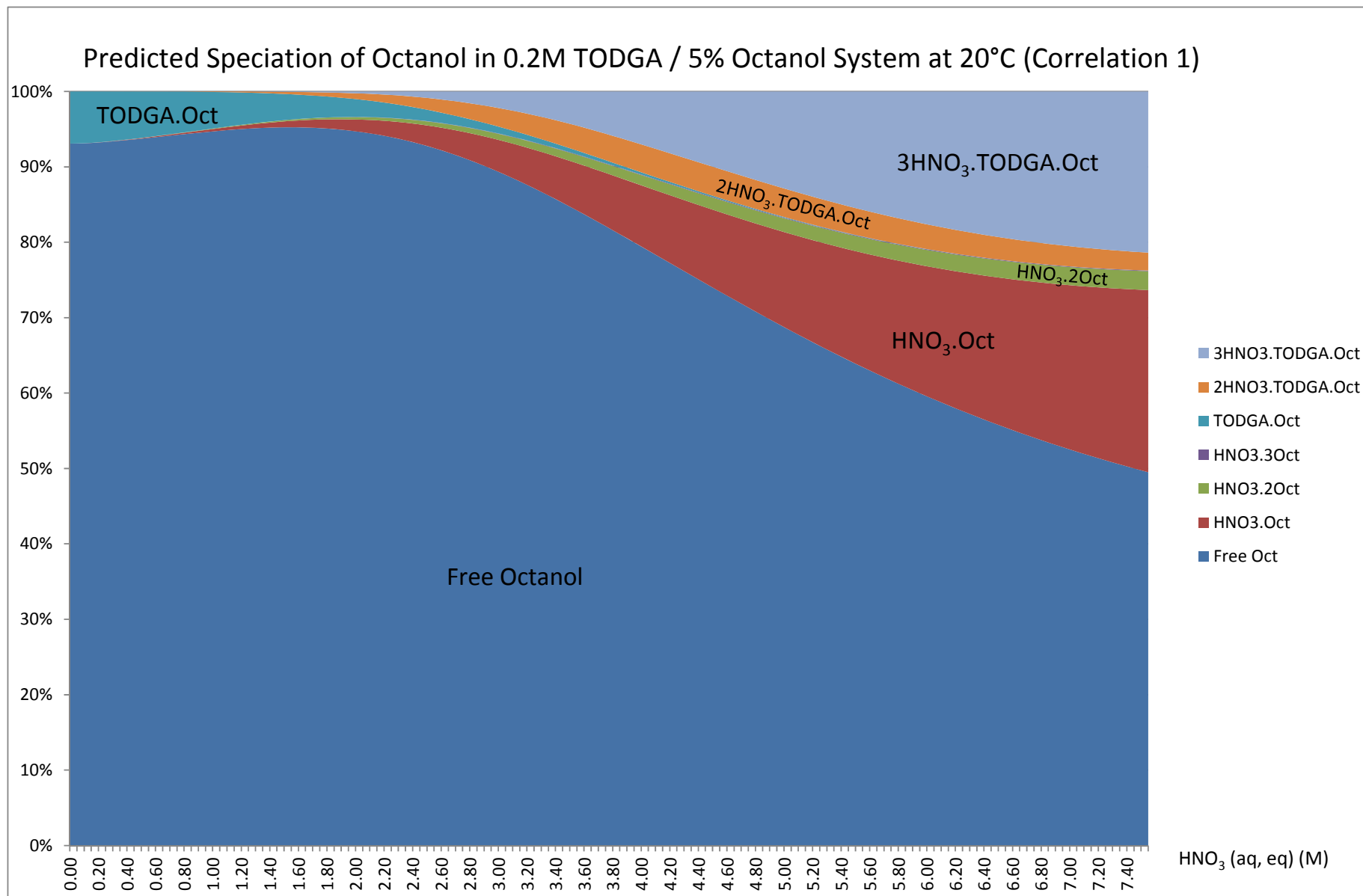


Residuals in Model with TODGA/Octanol Complexes

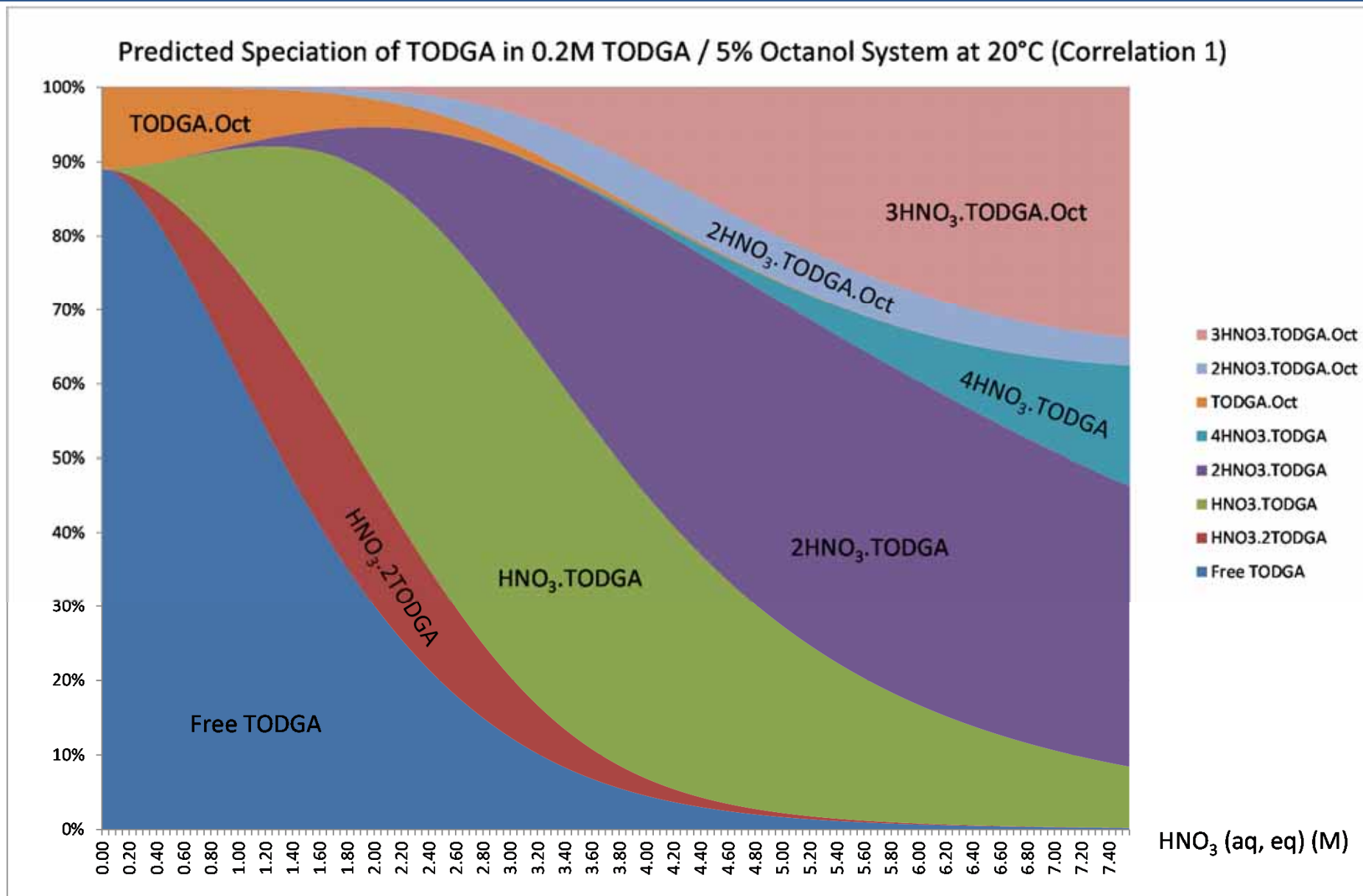
Mixed TODGA / Octanol Systems – Solvent Phase Acid Speciation



Mixed TODGA / Octanol Systems – Octanol Speciation



Mixed TODGA / Octanol Systems – TODGA Speciation



- Model is based on data in range 20 – 25°C
- Ideally have data for a wider range of temperatures so temperature dependence can be included.
- The model is based on data with acid as the only extracting species.
- Model will be required to work with metal loaded systems
 - Need to consider systems in which $[H^+] \neq [NO_3^-]$ implying need to distinguish $\gamma_{H^+}, \gamma_{NO_3^-}$
 - Some subtle issues related to dilution arise in systems with multiple extracting species

- Acid Extraction into Octanol / Diluent can be modelled to a high degree of accuracy assuming $\text{HNO}_3 \cdot n\text{Oct}$, $n = 1, 2, 3$
 - Acid Extraction into TODGA / Diluent can be modelled to good accuracy assuming $\text{HNO}_3 \cdot 2\text{TODGA}$ and $n\text{HNO}_3 \cdot \text{TODGA}$, $n = 1, 2, 4$.
 - Combining the two models suggests some synergistic and antagonistic effects
 - These can be incorporated in model by inclusion of $n\text{HNO}_3 \cdot \text{TODGA} \cdot \text{Oct}$ complexes, $n = 0, 2, 3$
 - Wide scatter of data at low acidity makes validation in this region difficult
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